

## Kphqt o cvkqp "Hqt o "hqt" ULVW" I t c f w c v g " Rtqhguukqp " Eqwtugu "

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* Course Name	Chinese			
	English Kinetics of Materials			
* Credits	3	* Teaching Hours	48 1 =16	
* Semester	Spring	* Cross-semester?	No	Spanning over Semesters
* Course Type	Program Core Course	* Course Type	For full-time students	
* Course Category	Specialized Course	Targeting Students	All graduates	
* Instruction Language	English	Teaching Method	In class teaching	
* Grade	Letter grading	Exam Method	Written Exam	
* School				
Subject				
Person in charge	Name	ID	School	E-mail
				hongwang2@sjtu.edu.cn
Gzvpgfgf "Kphqt o cvkqp"				
* ( ) Course Description	200			
	1) 2) 3) 4)			
* English Course Description	<u>This course provides a foundation for the advanced understanding of the phenomenological and atomistic kinetic process in materials. It emphasizes comprehension of fundamental concepts and stresses on development of students' ability of quantitative analysis. The course starts from a brief review of classical thermodynamics necessary for understanding of phase diagrams, followed by some key concepts such as flux and driving force. The principles of chemical reaction kinetics will be introduced, as well as several important applications such as in thin film growth process and chemical vapor deposition. The center stage of the course is given to the diffusion process in solid materials as well as the analytical and numerical methods to solve diffusion problems. Then</u>			



<p>* English Syllabus</p>	<p>The class will be delivered in 24 lectures, where each lecture contains two 45min classes.</p> <p><b>I INTRODUCTION AND BACKGROUND</b>  Introduction thermodynamics 1 lecture  Phase diagrams, Driving force, flux 1 lecture</p> <p><b>II KINETICS OF CHEMICAL REACTIONS</b>  Chemical reaction kinetics, adsorption isotherms 1 lecture  Thin film growth Rate controlling steps; CVD 1 lecture</p> <p><b>III DIFFUSION IN SOLIDS</b>  Fick's Laws and solutions to Fick's laws 2 lecture  Interdiffusion 1 lecture  Self, tracer, intrinsic and interdiffusion coefficients 1 lecture  Atomistic models of diffusion, Diffusion in ionic crystals 1 lecture  Multipath imperfections 1 lecture</p> <p>Midterm Exam</p> <p><b>IV KINETICS DRIVEN BY CAPILLARITY FORCES</b>  Capillarity forces on surfaces, grain growth 2 lectures  Surface energy anisotropy 1 lecture  Particle coarsening, sintering 1 lecture</p> <p><b>V KINETICS OF PHASE TRANSFORMATIONS</b>  Nucleation and growth 2 lectures  Solidification 1 lecture  Order-disorder Reactions 1 lecture  Spinodal decomposition 1 lecture  Martensitic transformation 1 lecture</p> <p><b>VI MODELLING OF KINETIC PROCESS</b>  Computational thermodynamics (CALPHAD) 1 lecture  Diffusion simulation, Phase field simulation 2 lecture  Computational lab 1 lecture</p> <p>Final Exam</p>
<p>* Requirements</p>	<p style="text-align: right;">50</p> <p>1.10 2% 20%</p> <p>2. 1 10 30%</p> <p>3. 30%</p> <p>4. 10%</p> <p>5. : 5%</p> <p>6. : 5%</p>
<p>* English Requirements</p>	<p>1. 10 Homework assignments, 2% each, 20%</p> <p>2. Midterm Exam, covers first 10 lectures 30% each</p> <p>3. Final Exam, covers later 9 lecture, 30%</p> <p>4. Participation: attendance and class discussion, 10%</p> <p>5. Quiz: simple questions due next morning 6:00am after each class, 5%</p> <p>6. Project: computation lab work, 5%</p> <p>The final grade will be calculated based on the sum of the above.</p>
<p>* Resources</p>	<p>_____ : KINETICS IN MATERIALS SCIENCE AND ENGINEERING Dennis W. Readey, CRC Press, 2017</p>

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